Optimized Corrosion Control Treatment: Bench-Scale Testing for Portland Water Bureau Damon K. Roth, PE, BCEE

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PNWS-AWWA 2018 Section Conference Tacoma, WA



### Acknowledgements







### **Overview**

- Why was this study needed? – Impact of changing DIC
- How was the study structured?
  Bench-scale test plan
- What were the results?
  - Pb release:
    - Existing conditions
    - GW Orthophosphate CCT
    - SW Orthophosphate CCT
    - SW pH/alkalinity CCT

### WHY WAS THIS STUDY NEEDED?



### **PWB Source Water**

- Surface water (Bull Run) pH ~8.0 Alkalinity ~7 mg/L as CaCO<sub>3</sub> (~1.5 mg/L as C)
- Groundwater (CSSWF) pH ~8.0 Alkalinity ~90 mg/L as CaCO<sub>3</sub> (~22 mg/L as C)
- System might be supplied with:
  - -100% SW
  - SW supplemented with GW
  - 100% GW

# Classic Lead Solubility as a Function of pH



Source: Adapted from Schock and Lytle 2011

# Improved Understanding of Lead Scales



- Cerrusite
  - $Pb(II)CO_3$  (s) lead carbonate
  - Less stable
  - More prone to sloughing, formation of particulate lead
  - Dissolves easily when WQ not favorable to production



- Hydrocerrusite
  - Pb(II)<sub>3</sub>(CO<sub>3</sub>)<sub>2</sub>(OH)<sub>2</sub> (s) anhydrous lead carbonate
  - More stable
  - Dissolves at higher DIC

### **Potential for Scale Change**





### **Potential for Scale Change**





### **Potential for Scale Change**





### **Bench-Scale Test Goals**

- Compare relative performance of CCT
  - Orthophosphate to pH/alkalinity
  - Orthophosphate doses
  - pH/alkalinity
- Evaluate potential for Pb release under changing water quality conditions
  - $-\,100\%$  SW to SW:GW blend
  - 100% GW to SW:GW blend

### HOW WAS THIS STUDY STRUCTURED?



### **Overview of Study and Methods** Test 1 - pH/Alkalinity Testing (Phase 1)



Source	SW	SW	SW	GW	
рН	8.0	8.6	9.3	8.0	
ALK	7	35	25	90	$(mg/L as CaCO_3)$
(	CU/PB	CU/PB	CU/PB	CU/PB	-
(	РВ	РВ	РВ	РВ	Replicates
[	BR	BR	BR	BR	

### **Overview of Study and Methods** Test 1 - pH/Alkalinity Testing (Phase 1)





### **Overview of Study and Methods** Test 2 – Orthophosphate Testing (Phase 1)



### **Overview of Study and Methods** Test 1 - pH/Alkalinity Testing (Phase 2)



### **Overview of Study and Methods** Test 2 – Orthophosphate Testing (Phase 2)



### WHAT WERE THE RESULTS?





#### **1. EXISTING CONDITIONS**

#### 2. GW EQUILIBRATED ORTHOPHOSPHATE

#### 3. SW EQUILIBRATED ORTHOPHOSPHATE

#### 4. SW PH/ALKALINITY ADJUST

### **Existing Conditions** Lead Coupons – All Blends



### **Existing Conditions** Copper Pipes (Lead Solder) – All Blends





#### **1. EXISTING CONDITIONS**

#### 2. GW EQUILIBRATED ORTHOPHOSPHATE

#### 3. SW EQUILIBRATED ORTHOPHOSPHATE

#### 4. SW PH/ALKALINITY ADJUST

#### **GW Lead Coupons with Orthophosphate** 15% Surface Water Blend



#### **GW Lead Coupons with Orthophosphate** 85% Surface Water Blend



# GW Copper Pipes (Pb Solder) with Orthophosphate 85% Surface Water Blend





#### **1. EXISTING CONDITIONS**

#### 2. GW EQUILIBRATED ORTHOPHOSPHATE

#### **3. SW EQUILIBRATED ORTHOPHOSPHATE**

#### 4. SW PH/ALKALINITY ADJUST

#### SW Lead Coupons with Orthophosphate 85% Groundwater Blend



# SW Copper Pipes (Pb Solder) with Orthophosphate 85% Groundwater Blend





#### **1. EXISTING CONDITIONS**

#### 2. GW EQUILIBRATED ORTHOPHOSPHATE

#### 3. SW EQUILIBRATED ORTHOPHOSPHATE

#### 4. SW PH/ALKALINITY ADJUST

# pH Stability Testing

- Phases 1 and 2 indicated pH was not stable in test reactors with alkalinity at 25 mg/L as CaCO<sub>3</sub>
- New testing evaluated stability of pH 9.3 and pH 9.5 at 30, 34, 40, and 44 mg/L as CaCO<sub>3</sub>
  - Samples with and without headspace were also evaluated



## Post-Stagnation pH Values, Without Headspace During Coupon Study



### **CCT Comparison – Lead Coupons**



### **CCT Comparison – Brass Coupons**



Normalized Lead Concentration (µg/L-day)

# **CCT Comparison – Copper with Lead Solder Reactors**



# **Additional Blending Testing**

- Coupons equilibrated with GW were exposed to a blend of GW + SW at pH 9.3
   Blend ratios were 85:15 and 15:85 SW:GW
- Coupons equilibrated with SW treated with 0.75 mg/L as P were exposed to a blend of SW with 0.75 P + GW without P

– Blend ratios were 20% and 50% GW

# Lead Coupons GW + SW at pH 9.3



## Brass Coupons GW + SW at pH 9.3



# Copper with Lead Solder Reactors GW + SW at pH 9.3 (target)



# Lead Coupons SW with 0.75 P + GW with 0.0 P

→ 80% SW; 20% GW - ◆ 50% SW; 50% GW



### Brass Coupons SW with 0.75 P + GW with 0.0 P

→ 80% SW; 20% GW • → 50% SW; 50% GW



## Copper with Lead Solder Reactors SW with 0.75 P + GW with 0.0 P







### Summary

- Both orthophosphate and pH/alkalinity reduced Pb release
  - Ortho CCT had lowest lead levels in lead coupons
  - High pH/alkalinity CCT had lead levels equal to ortho for brass coupons
  - pH could not be maintained in the lab for the copper/lead solder coupons
- If high pH is used for CCT important to maintain pH in system

## Conclusions

- Condition with most potential for Pb release is GW to SW (cerussite to hydrocerussite)
  - Can be reduced or eliminated with CCT (orthophosphate or pH/alkalinity)
  - This should be monitored in the system if this method is selected
- When PWB GW without ortho was blended into coupons exposed to SW + ortho, the data did not show an increase in lead after blending in GW



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